

February 11, 2021
MSE 2103 topics

- The math behind radioactive decay – part 2
 - ^{137}Cs has a half-life of 30.17 years as it decays into ^{137}Ba . How much Cs is left after 50 years? How much Ba is produced after 75 years?
 - ^{238}U has a half-life of ~4.5 billion years. A rock from lunar highlands shows that 45% of the original uranium decayed into lead. How old is that rock?
 - Determine the half-life of ^{40}K if you know that there is ~9% of primordial potassium left in an asteroid that is ~4.5 billion years old.
- Lines of evidence for a 4.6 billion year old Earth
- Fine, Earth is 4.6 billion years old; but what about the Sun?
 - chemical reactions, fossil fuel burning?
 - radioactive decay?
 - gravitational contraction?
 - ... or something else?
- $E = mc^2$
- Eddington's model (from Einstein's mass-energy equivalence)
- The proton-proton chain: $p+p \rightarrow p+n \rightarrow p+p+n *2 \rightarrow \text{He-4}$
- ... except... how do protons overcome extreme requirements?